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# Effect of austenitizing condition on carbon content and volume fraction of austenite of an alloyed ADI

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#### الملخص

تم في هذا البحث در اسة عينات من الحديد المطاوع تحتوى على عناصر سبائكية ( نحاس و نيكل بنسبة 1.6% لكلا منهما) خضعت لمعالجة حرارية تدعى (austempering) عند درجة حرارة ثابتة 360°م و لزمن ثابت 180 دقيقة. بعد معالجة حر ارية تدعى (austenitizing) للوصول بالسبيكة لطور الأوستينيت لدرجتي حرارة مختلفتين (900, 1000°م) و لأزمان مختلفة أيضا. مبتدئا من 15 دقيقة, 60, 120, و منتهيا بـ 180 دقيقة. و در استنا تهتم بتأثير ظروف المعالجة (الحرارة و الزمن) على المحتوى الكربوني C<sub>γ</sub> و كذلك تأثيرها على كمية الأوستينيتX حاصل ضرب كمية الأوستينيت و محتواها الكربوني X<sub>Y</sub>C<sub>Y</sub> و أخيرا على حجم الفيرايت الأبري. و باستخدام حيود الأشعة السينية لتقدير كلا من: كمية الأوستينيت X<sub>v</sub> و نسبة الكربون بداخله/C و كذلك حاصل ضربهما (XyCy) بأتباع خطوات طريقة كلا من Rundman و Klug و أخيرا حساب حجم حبيبات الفير ايت باستخدام Scherrer formula. لوحظ من خلال النتائج أن المحتوى الكربوني داخل الأوستينيت يزداد بزيادة زمن التسخين لطور الأوستينيت و أن كمية الأوستينيت المتبقى و الثابت عند درجة حرارة الغرفة بزداد بزيادة كلا من الزمن و درجة الحرارة. أما عن تأثير المعالجة الحرارية (austenitizing) على التركيب المجهر ي للعينات فقد تم در استه و لوحظ أن هناك تغير ملحوظ في شكل و حجم الفير ايت الأبري و كذلك مظهر الأوستينيت المتبقى. حيث أن الأوستينيت يظهر على شكل رقائق فضية متداخلة بين الفيرايت الأبرى وكذلك بمظهر كتلى أيضا. وأن أفضل زمن لعملية (austenitizing) و التي تتطابق مع أفضل تركيب مجهري مرغوب فبه عند 180 ثم 120 دقبقة.

#### Abstract

A ductile iron containing 1.6 wt% Cu and 1.6 wt% Ni as the main alloying elements was austempered at a fixed austempering temperature of 360°C for a fixed austempering time of 180 min after austenitizing at 900°C for different austenitizing periods of 15, 30, 60, 120, and 180 min. the austempering process was repeated after changing austenitizing temperature to 1000°C. The effect of austenitizing temperature and time was studied on structural parameters like the carbon content in austenite  $C_{\gamma}$  after austenitizing, the volume fraction of austenite after austenitizing  $X_{\gamma}$ , their product  $X_{\gamma}C_{\gamma}$ , and the ferrite needle size  $d_{\alpha}$  after austempering.

X-ray diffraction analysis was performed to estimate the percentage of carbon in austenite  $C_{\gamma}$ , the austenite content  $X_{\gamma}$ , following the procedure of Rundman and Klug, and so the particle size  $d_{\alpha}$ , have been investigated by using the Scherrer formula.

The carbon content in austenite increases with increasing the austenitizing time, and so the amount of retained austenite increases with increasing both times and temperatures. The effect of austenitizing parameters were also studied on austempered microstrucure.

The qualitative variation in morphology, size of ferrite needles and retained austenite in austempered microstructure with austenitizing conditions were observed.

The austenitizing conditions not only affect the morphology of ferrite but the retained austenite. Austenite appears to exist in the form of films between ferrite needles, as well as in blocky morphology.

The austenitization time corresponding to most desirable austempered microstructure is 180 followed by 120 min.

#### **<u>1- Introduction</u>**

Austempered ductile cast iron (ADI) has a better combination of strength, ductility, toughness, fatigue and wear resistance. Thus, it is considered an important engineering material, that is useful in a great number of applications. Many applications of ADI have been reported earlier [1-4]. ADI is obtained by subjecting the ductile cast iron to austempering heat-treatment cycle. The austempering reactions in ADI occur in two stages [5]. In stageI, the matrix austenite with carbon content  $C^{\circ}_{\gamma}$ isothermally transforms into ausferrite, i.e., into a mixture of acicular ferrite with carbon enriched stabilized austenite with carbon content  $C_{\gamma}$ %. In stageII, the stabilized austenite decomposes into ferrite and carbide. Previous investigators [6-8] have studied the effects of austenitising temperature  $(T_{\gamma})$ , austempering temperature  $(T_A)$  and austempering time  $(t_A)$  on the structure and mechanical properties of this material but very little work has been carried out on the influence of austenitising time  $(t_{\gamma})$ . Thus this work studies the austenitizing transformations and microstructural changes during austempering of alloyed ductile iron. Data from the literature have been analyzed to understand aspect of the retained austenite in austempered ductile irons. The current study deals with the effect of austenitization temperature and time on the average carbon content and volume fraction of austenite after austempering.

## **<u>2- Experimental Procedure</u>**

# 2.1- Material and Heat treatment

The ductile cast iron (DI) used in the present work has the following chemical composition (wt. %): 3.3C, 2.6 Si, 0.35Mn, 0.008S, 0.01P. 0.05Mg, 1.6Cu, 1.6Ni, balance Fe. It was supplied in the form of round cylindrical (20mm) diameter and 5mm long specimens made of this DI were subjected to cycle heat treatment to produce austempered ductile cast iron: austenitizing at 900°C and 1000°C for different times at each temperature (15, 30, 60, 120, and 180 min), then austempering at 360°C for 180 min in a salt bath with subsequent cooling to ambient temperature to prevent any transformations of austenite into pearlite. The resulting microstructure contained a ferritic matrix with some spheroidized carbides and graphite spheres. The latter have nodularity in excess of 90% and an average diameter of 0.045 mm. The graphite nodules represent a volume fraction of 12% and have an average count of  $75/\text{mm}^2$ .

# **2.2- Microscopy and X-Ray Analysis**

Specimens were polished by using standard metallographic techniques and etched with the 2% nital solution. Light optical microscope (Nikon, FX-35DX) was used for microstrucural

analysis. The phases and constituents of the matrix were identified by using Cu-K<sub> $\alpha$ </sub> radiation with Ni filter at 40,000 volt difference between a tungsten filament and copper target 30mA. X-Ray diffraction (XRD) analysis was performed to estimate the austenite content, the percentage of carbon in austenite, and their product X<sub> $\gamma$ </sub>C<sub> $\gamma$ </sub> following the procedure of Rundman and Klug. And so the particle size d<sub> $\alpha$ </sub>, have been investigated by using the Scherrer formula. The XRD equipment used in this work is Philips–binary (scan) of a PW.1800. A diffraction pattern records the X-ray intensity as a function of 2-theta angle. All the diffraction patterns were prepared as step-scans. To run a step-scan set the tube voltage and current, and enter the following parameters should be entered:

Start Position [2°Th.]2.0100End Position [2°Th.]99.9900Step Size [2°Th.]0.0200Scan Step Time [s]0.5000

Once started, the goniometer moves through its range, stopping at each step for the allotted time. The X-ray counts at each step are saved to a file on the computer. The profiles were analyzed on a computer to obtain the peak positions and the integrated intensity of (111), (220) and (311) peaks of austenite and (110) and (211) peaks of ferrite. The volume fraction of ferrite  $X_{\alpha}$  and austenite  $X_{\gamma}$  were determined by using the direct comparison method for a mixture of ferrite and austenite, the direct comparison expression can be written as[7]:

$$\frac{I_{\gamma}}{I_{\alpha}} = \frac{R_{\gamma} X_{\gamma}}{R_{\alpha} X_{\alpha}} \tag{1}$$

Where  $I_{\gamma}$  is the integrated intensity from a given reflection plane of austenite phase  $\gamma$ ;  $I_{\alpha}$  is the integrated intensity from a given plane of ferrite  $\alpha$  phase and  $R_{\gamma}$  and  $R_{\alpha}$  are constants obtained from following R value expression [7]:

$$R=1/v^{2} (|F|^{2} pL) e^{-2M}$$
 (2)

Where v is the volume of unit cell; *F* is the structure factor; *p* is the multiplicity factor; *L* is the Lorentz-polarization factor; and (e<sup>-2M</sup>) is the temperature factor. The quantity M depends on the amplitude of thermal vibration and scattering angle. The carbon content of retained austenite calculated according to angular positions of the FCC austenite peaks,  $C_{\gamma}$  can be calculated from equation (3):

$$C_{\gamma} = (a_{\gamma} - 3.555) / 0.044 \tag{3}$$

Hence  $a_{\gamma} = 0.3555 + 0.0044 C_{\gamma}$ , where  $a_{\gamma}$  is the lattice parameter of austenite in last equation in nano meter [8]. An estimate of ferritic cell size (d<sub>\alpha</sub>) obtained from the x-ray diffraction profile from the breadth of the {211} diffractometer peaks of ferrite using the Scherrer formula [9].

$$d_{\alpha} = \frac{0.9\lambda}{\beta\cos\theta} \tag{4}$$

Where  $\lambda$  is the wave length,  $\beta$  is the breadth of the (211) peak of ferrite at half height in radians and  $\theta$  is the Bragg angle.

<u>3- Discussion of Results</u> a- The effect of austenitization temperature on the microstructure has been studied by comparing the micrographs in figures (1) and (2) for similar

austenitization times of 15, 30, 60, 120, and 180 min. The results indicate that the increase in the austenitization temperature from 900-1000°C influences the austempered microstructure in various ways, the austempered microstructure has coarsened, the volume fraction of retained austenite has increased, and the amount of blocky retained austenite has detected.



(a)

(b)



(d)



(e)

Figure (1) Microstructures of an alloyed ADI samples at  $T_{\gamma}$ =900°C, for different  $t_{\gamma}$ : (a) 15, (b) 30, (c) 60, (d) 120, and (e) 180min, 15



(a)

(b)



(c)



(e)

Figure (2) Microstructures of an alloyed ADI samples at T<sub>r</sub>=1000°C, for different t<sub>r</sub>: (a) 15, (b) 30, (c) 60, (d) 120, and (e) 180min, 600x.

b- The austempered microstructure of ADI formed by austenitizing at 1000°C for 15min consists of acicular ferrite in the form of fine needles distributed more uniformly throughout the structure, with the retained austenite appearing as silvers in between these ferrite needles. At higher austenitizing temperature and lower time, the nucleation of ferrite may be expected to be faster, but the carbon rejected during formation of ausferrite may diffuse only to shorter

distances thereby forming fine ferrite and austenite. During austenitization, the cast microstructure transforms to austenite. and its carbon content depends upon the initial microstructure including nodule size and nodule count, austenitization temperature, and time. In a fully ferritic matrix, carbon diffuses only from the graphite nodules to the surrounding matrix during austenitization. The graphite nodules are the only source of carbon, and consequently, the carbon diffusion distances involved during solution treatment may be relatively large. This is nearly true because very little carbon can be attained from the small quantities of spheroidized carbides present. Consequently, full austenitization requires either very long solution treatment cycles or a very high carbon diffusion rate. However, in a pearlitic matrix, the graphite nodule and cementite of pearlite contribute to the carbon enrichment of the mother austenite, which is subsequently subjected to austempering at a given temperature, taking place in two stages. Figure (1) shows the variation of austempered microstructure with austenitization at 900°C for time periods 15, 30, 60, 120 and 180 min. The austempered microstructures consist of acicular ferrite and retained austenite but it is nonuniform when austempered after 15, 30 min of austenitization at 900°C. As the time of austenitization is increased to 60, 120, and 180 min, the acicular ferrite and retained austenite get distributed more uniformly due to homogeneous carbon content of austenite in the matrix. However, for the same austempering treatment after austnitization at 1000°C for different times, the microstructures are significantly different from those austenitization at 900°C, figure (2). The austempered structure is uniform for 180 min of austenitization, as the carbon content has become uniform after 180 min due to higher diffusion coefficient at 900, 1000°C, as the time of austenitization is increased, the

resulting austempered microstructure becomes coarser, which may be attributed to grain growth of austenite resulting in lower rate of heterogeneous nucleation of ferrite. Thus, a coarse structure results with retained austenite present as large blocky areas. The number of ferrite platelets is much greater for the samples austenitized at 900°C then those at 1000°C for the same austempering condition, of 360°C for 180 min as observed in figure (1), and (2). As described former, the XRD measurements allowed estimates to be made of the austenite lattice parameter, and then the austenite carbon content of each sample of ADI, figure (3) shows how the carbon content of the retained austenite varied with austenitizing conditions.



Figure (3) Effect of austenitizing conditions on carbon content of austenite in ADI samples

This agreed with figure (4) which show the schematic free energy- composition diagram [10], thus at lower austenitizing temperature (900°C), a greater driving force can affect the rate of stage I in two ways: the number of ferrite nuclei may increase directly and / or the activity gradient driving carbon diffusion may increase, this is happened. Lowers the driving force controlling the transformation of austenite to ausferrite from ab to a'b', the driving force reduction is responsible for the decrease in the number of ferrite nuclei formed and the slower growth along the ferrite platelet. Therefore, as shown in figure (2) increasing the austenitizing temperature to 1000°C leads to structures containing a high percentage of large austenite grains.



Figure (4) Schematic free energy- composition diagram indicating the driving force for stage [I] and stage [II] of austempering [10].

**c-** Another results obtained by calculating volume fraction of austenite by using direct comparison method show that: increasing both times and temperatures of austenitizing followed by increasing in austenite content figure (5) shows a plot of the volume fraction of austenite against the austenitizing conditions. The austenite content was observed to increase with increasing austenitizing temperatures.



Figure (5) Effect of austenitizing conditions on volume fraction of austenite calculated by direct comparison method.

**d-** A Computer program employed at the Libyan petroleum Institute is used to carry out the measurements for quantitative analysis automatically for each sample, as is shown in figure (6).



Figure (6) Effect of austenitizing conditions on volume fraction measured by diffractometer.

e- From traditional data, the yield strength of ADI depends on the fineness of ferrite and austenite, therefore the fracture toughness, fatigue strength of ADI have been observed to depend on the parameter (austenitic carbon content,  $X_{\gamma}C_{\gamma}$ ). As shown in figure (7) where the austenitic carbon content ( $X_{\gamma}C_{\gamma}$ ), i.e. the total carbon in austenite is plotted against the austenitizing time.



Figure (7) Effect of austenitizing conditions on austenitic carbon (calculated & measured).

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**f**- Figure (8)shows the variation of the ferritic cell size  $(d_{\alpha})$  with the austenitizing time at 900°C.



Figure (8) Influence of austenitizing time on ferritic cell size at 900°C.

#### 4- Conclusions

The following conclusions can be drawn:

- The amount of the carbon content in austenite increases with increasing austenitizing time, and retained austenite increases with increasing both the time and temperatures of austenitization.
- The volume fraction of retained austenite X<sub>γ</sub>, its carbon content C<sub>γ</sub>, and X<sub>γ</sub>C<sub>γ</sub> are small at short austenitizing time.
- In a ferritic matrix structure, significantly more time is required to reach equilibrium carbon content in the mother austenite. This is attributed to the larger carbon diffusion distances and the presence of copper, which segregates to the graphite- metal interface and creates a carbon diffusion barrier.
- The austenitizing conditions not only affect the morphology of ferrite but also that of retained austenite. Austenite appears to exist in the form of films between ferrite needles, as well as in blocky morphology.
- In ADI austenitizing at higher temperatures, it has coarser or feathery ferrite and austenite and by last

knowledge, this reduces the yield and tensile strengths but imparts higher ductility.

- Increasing T<sub>γ</sub> from 900 to 1000°C decreases the free energy controlling the transformation of austenite to ferrite and high carbon austenite. Less ferrite nuclei are formed at 1000°C, and the resulting structure is coarser and contains more blocky type austenite.
- The austenitizing time corresponding to most desirable austempered microstructure is 180 followed by 120 min.
- The austenitizing treatments exert well-pronounced effects upon the microstructure, then mechanical properties of austempered ductile iron.
- Previous results [11] indicated that the austenitizing temperature T<sub>γ</sub>, and time t<sub>γ</sub>, have a significant effect on the mechanical properties for the same alloys, this can be attributed to the influence of these variables on the carbon kinetics.

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### Influence of Design Tip Speed Ratio and Rated Wind Speed on Energy Yield of Horizontal Wind Turbine

### Mohamed Ali Naili

**Abstract** Wind turbines are the means of converting wind energy into electrical energy. It is not unexpected that certain brands of horizontal wind turbines are optimally designed for certain wind speed regimes prevailing at certain sites. It has been the objective of this study to find out whether it is possible to design a wind turbine that is more suitable for a site of lower annual mean wind speed such as the western coast of Libya. In achieving this objective, use was made of a typical wind turbine aerodynamic design and performance analysis procedure together with actual wind speed data recorded at the city of Misurata.

Based on the annual wind energy yield obtained from different designs of wind turbines having different rated wind speeds and different blade design tip speed ratios, this study indicates that the decrease of the rated wind speed leads to a continuous increase of annual energy yield as well as an increase of the cost of the wind turbine. Therefore the optimum value of rated wind speed for this site may only be determined as a compromise between added wind energy yield and added cost. Moreover, for any given rated wind speed this study indicates that a blade design tip speed ratio equal to six seems to be the optimum value for the given site.

**Objective** It is not unexpected that certain brands of horizontal wind turbines are optimally designed for a certain wind speed regime or regimes prevailing at a certain areas of the world. In other words, the wind turbine is designed such that it would work most efficiently if it were installed at a site of a given wind speed regime. On the other hand it would work less efficiently if it were installed at a site of a different wind speed regime. It is

thus expected that several wind turbines that are made in Europe were designed to suit the wind speed regimes prevailing in that area. Since the annual mean wind speed in Europe is relatively higher than the corresponding value in Libya, it is not unexpected that European-made wind turbines would work less efficiently here.

The objective of this study is to find out whether it is possible to design a wind turbine that is most suitable for a wind regime of lower annual mean wind speed. In achieving this objective, usage was made of a typical wind turbine aerodynamic design procedure together with actual wind speed data of a certain Libyan site. Analysis and comparison of the final results were based on the annual wind energy yield at the site obtained from the different wind turbines.

**The P(V) curve** The power output of the wind turbine varies with the wind speed as shown schematically in Figure (1). It is noted from this figure that the wind turbine does not produce any power until the wind speed, V, exceeds what is termed the 'cut-in wind speed', V<sub>c</sub>. Beyond this velocity the power output, P, increases as V increases. When the wind speed reaches a value termed the 'rated wind speed', V<sub>r</sub>, the wind turbine produces its rated power output, P<sub>r</sub>. From thereon, as the wind speed increases the power output remains constant. Once V reaches a value termed the 'furling wind speed', V<sub>f</sub>, the value of P becomes zero [6].

It is expected that wind turbines commercially available in Europe, for example, are more suitable for such a continent where wind speed patterns show relatively higher annual average wind speeds. It is well known that the annual average wind speed in Libya is relatively low, so it is not unexpected that the performance of those turbines in Libya will be less efficient than if operated in Europe. The P(V) curve is constructed from the  $C_P(\lambda_0)$  curve of the wind rotor plus the P( $\omega$ ) of the electrical

generator to be employed, where ' $\omega$ ' represents the rotational speed of the generator. The wind turbine power curve strongly depends on the wind rotor  $C_P(\lambda_0)$  curve.



Fig. (1): Schematic drawing of typical P(V) curve [8].

**Analytic procedure** In order to determine the annual energy yield at the selected site for any of the case studies, a sequence of computational procedures were followed. This sequence consisted of wind rotor aerodynamic design followed by performance analysis of the designed rotor and then the construction of the power curve for the resulting wind turbine.

#### Wind energy yield at a site

**1. Wind speed data** For the sake of this study, raw wind speed data recorded by Libyan National Meteorological center at the city of Misurata were employed. This data set contains measurements of mean wind speed at 3-hours intervals taken at a height of 10 meters above ground level. The data was recorded in 2010 and covers a period of 12 months. It is found that mean wind speed for this site was 5.3 m/s.

### 2. Method of evaluation

By definition, the power P is given by:

 Hence, E =  $\int P(t)dt$ .....(2)

Now given the power curve of the wind turbine P(v), and given the raw wind data V(t), the integral in (2) above may thus be evaluated as follows

$$E \approx \sum_{i=1}^{i=N} Pi(Vi) \Delta t$$

.....(3)

Where:

N = total number of wind speed data points.

 $\Delta t = 3$  hours in this case.

 $P_i(V_i)$  = electrical power output when V=V<sub>i</sub>

The units of E would be (KW-hr), since P is evaluated in units of (KW).

Since N is large, the summation in (3) would be a good approximation to the integral given by (2). Moreover, since the proper evaluation of energy yield at the selected site is not the objective of this study, this approximation is thus justified.

**Description of cases** Table (1) below shows a total of thirty cases studied with a symbol assigned to each one. The letter 'T' stands for the design tip speed ratio while 'V' stands for the rated wind speed. So for example the symbol 'T5V11' refers to the case in which  $\lambda_{0D}$  and V<sub>r</sub> were equal 5 and 11 respectively. Normally, values of  $\lambda_{0D}$  of around 7 are used in commercially available wind turbines. However, for the purposes of this study lower values of  $\lambda_{0D}$  were investigated, namely 6, 5 and 4 as shown in table (1). Moreover, it is noticed that values of rated wind speed of a substantial number of existing horizontal axis wind turbines lie in a range of 12 to 14 (m/s). Once again, lower values of  $V_r$ , namely 11 and 10 were investigated in order to study their effect on energy yield at sites of generally lower annual mean wind speeds.

Table (1): Cases studied with their respective symbols

V <sub>r</sub>		$\lambda_{0D}$				
(11/8)	4	5	6	7	8	
10	T4V10	T5V10	T6V10	T7V10	T8V10	
11	T4V11	T5V11	T6V11	T7V11	T8V11	
12	T4V12	T5V12	T6V12	T7V12	T8V12	
13	T4V13	T5V13	T6V13	T7V13	T8V13	
14	T4V14	T5V14	T6V14	T7V14	T8V14	
15	T4V15	T5V15	T6V15	T7V15	T8V15	

Wind rotor design For the given values of  $\rho$ ,  $\eta_{GB}$ ,  $\eta_{EG}$ ,  $P_r$ ,  $Cp_r$ , the radius of the rotor, R, may be determined for any given value of the rated wind speed,  $V_r$ , using the mathematical relationship.

$$P_r = \frac{1}{2} \rho C p_r \pi R^2 V_r^3 \eta_{GB} \eta_{EG}$$

Since the rotor diameter varies inversely with the square of rated wind speed, the blade chord distribution for any given design tip speed ratio is determined in non-dimensional form, namely as C(r/R). In this manner, for a given value of design tip speed ratio one obtains a group of "geometrically similar blades" though with different actual values of rotor radii and actual values of chord distributions. Consequently, the Cp( $\lambda_0$ ) curves for the whole group (with the same  $\lambda_{0D}$ ) are absolutely identical.



**Fig.** (2): Influence of  $\lambda_{0D}$  on blade chord distribution.

Wind rotor performance.  $Cp(\lambda_0)$  curves for the five wind rotor geometries were obtained as shown in Fig.(4). In performing the above procedure, use was made of the aerodynamic data contained which presents variation of lift and drag force coefficients as functions of angle of attack for the airfoil used, namely NACA 0012.

As can be seen from Fig.(4) the  $C_p(\lambda_0)$  curves follow the normal trend. The power coefficient increases with the increase of tip speed ratio and reaches a maximum value ranging from 0.46 to 0.475 and then falls down continuously. It can be seenthat the value of  $\lambda_0$  at which Cp attains its maximum value increases as  $\lambda_{0D}$  increases. This is mainly due to the fact that the rotor design is based on maximizing the local value of Cp at the selected design tip speed ratio. It is also noticed that the operational range increases slightly with the increase of  $\lambda_{0D}$ .



Fig.(3): Influence of  $\lambda_{0D}$  on twist angle distribution.

**Wind turbine power curve** The wind turbine power curve was constructed via the matching of the wind rotor power characteristics with nominal electrical generator power characteristics.

For each one of the five distinct rotor geometries, six different rotor diameters were employed corresponding to the six different rated wind speeds selected, namely 10, 11, 12, 13, 14 and 15 (m/s). In this manner six different wind turbine power curves were obtained for each of the five values of  $\lambda_{0D}$ . It is noted that in order to obtain any one of these power curves, the same nominal electrical generator

power characteristics were employed. The same results are presented in the form of power curves for a given value of  $V_r$  and different values of  $\lambda_{0D}$ .



Fig. (4):Variation of power coefficient with changes in tip speed ratio.

**Wind energy yield.** In order to investigate the effect of rated wind speed as well as the effect of rotor design tip speed ratio on energy yield, raw wind speed data collected at the city of Misurata were employed.

#### 1. Effect of rated wind speed

Fig. (5) presents curves of annual wind energy yield,  $E_{an}$ , as a f unction of rated wind speed for given values of  $\lambda_{0D}$ . It can be seen from this figure that the increase of  $V_r$  leads to a continuous decrease of  $E_{an}$  for all cases of  $\lambda_{0D}$ . This result is due to the shift of the power curve to the right as Vr increases. However, it must be noted that though decreasing the rated wind speed leads to increased wind energy yield, the substantial increase in the rotor diameter would certainly lead to increased cost of the rotor itself plus the cost of other wind turbine parts such as the tower and the mechanical parts of the control system. Therefore, to determine the optimum rated wind speed for a site, one has to analyze the added costs resulting from the reduction of Vr and weigh them against the gains in wind energy yield over the assumed life time of the wind turbine.

### 2. Effect of design tip speed ratio

Fig.(6) presents curves of  $E_{an}$ , as a function of  $\lambda_{0D}$  for given values of  $V_r$ . It can be seen from this figure that for any given value of rated wind speed, the annual energy yield increases with the increase of  $\lambda_{0D}$  up to a value of around 6. From thereon, the value of  $E_{an}$  decreases continuously. This is mainly due to manner in which the power curve changes with  $\lambda_{0D}$  for any fixed value of Vr. Thus for the given site, there seems to be an optimum design tip speed ratio of around 6 irrespective of the value of the rated wind speed.



Fig. (5): Influence of  $\lambda_{0D}$  on annual energy yield for different values of  $V_{\rm r}.$ 



Fig. (6): Influence of V<sub>r</sub> on annual energy yield for different values of  $\lambda_{0D}$ .

**Conclusions** From the foregoing discussion, the results of this study show that within the ranges of rated wind speed and blade design tip speed ratio explored and based on the analytical method employed the following conclusions may be drawn:

 $\div$ 

or the site employed in this study, the annual energy yield reaches a maximum at a value of design tip speed

F

ratio of around 6 irrespective of the value of the rated wind speed. This indicates that the optimum value of  $\lambda_{0D}$  for this site could well be equal to 6.

Т

S

G

S

Μ

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he decrease of rated wind speed leads to a continuous increase of annual energy yield as well as rotor radius for any given blade design tip speed ratio. This implies that no optimum value of  $V_r$  may thus easily be identified.

\*

ince decreasing  $V_r$  would lead to an increase in rotor radius which in turn would lead to an increase in the cost of the wind turbine, therefore the optimum value of  $V_r$  for a given site may only be determined as a compromise between added annual energy yield and added wind turbine cost.

From the foregoing statements it is clear that in order to determine whether it is possible to design a wind turbine that is most suitable for a wind regime of lower annual mean wind speed one has to supplement the results of the current study with corresponding results of another study that primarily deals with cost analysis resulting from changes of rotor diameter. Such a study is well beyond the scope of this thesis.

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# Slow release of entrapped *Azospirillum brasilense* cells

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#### Abstract

In this study *Azospirillum brasilense* st.was formulated in alginate pellets to study the slow release and survival of bacteria from alginate beads. Recorded in this study high number of *A. brasilense* cells were released from alginate pellets when starch was used as adjuvant, and recorded  $38.5 \times 10^4$  and  $10.7 \times 10^4$  cfu/g bead at initial and 8<sup>th</sup> day of sampling, respectively and The least cell number of *Azospirillum* was released by immobilization in alginate alone without any addition, recording  $27.5 \times 10^4$ - $13.1 \times 10^3$  cfu/g bead at initial and 8<sup>th</sup> day of sampling, respectively. Keywords: *Azospirillum brasilense*, survival, alginate pellets

#### **1.INTERODUCTION**

The encapsulation of cells methods was the best methods and most safe introduce Azospirillum in the soil where they are protected from environmental conditions and organisms competitors and increase their persistence and survival in the soil. (Kumaresan and Reetha, 2012). Azospirillum bacteria as plant growth promoting bacteria PGPB, where it has a positive impact on plant growth through its impact on root through nitrogen fixation and production of phytohormones and their survival depends on the soil on abiotic and biotic factors (Van Veen et al., 1997). the PGPB population in rhizosphere was inhibited by the decreased in the number of inoculated bacteria (Bashan, 1998). carrier bacteria on the gel material is one of the most successful methods for the introduction of bacteria in the soil (Vassilev et al., 2001). The primary goal of encapsulation PGPB the protection of vital and nonvital factors and RELEASE PGPB gradually to the soil to be able to colonize the roots (Bashan et al., 2002; Vassilev et al., 2001; El-Komy, 2001; 2005 ). and abiotic stresses such as the inhibitory effect of toxic compounds (Smit et al., 1996; Cassidy et al., 1997), to support the survival and activity Physiological (Weir et al., 1995; Trevors et al., 1993), In the present study, experiments were conducted to develop the gel based formulation of *Azospirillum brasilense* bioinoculant by enriched with different additives *viz.*, skimmed milk powder and starch

#### 2.MATERIALS AND METHODS

**Microorganisms** 

*A. brasilense* isolate has been obtained from free soil in Hai Al-salam district of Derna, Libya

Immobilization of Azospirillum brasilense isolate in alginate pellets:

A. brasilense isolate was immobilized by entrapment in 2% Ca-alginate. Cells encapsulated in alginate pellets were prepared by using the method applied at our laboratory (Shaban and El-Komy, 2000; El-Komy, 2001; 2005). Briefly, 25 ml of bacteria cell suspension under aseptic condition were added to75 ml of sterile alginate solution to obtain a final concentration of 2% alginate. In some cases skim- milk (2.5% w.v) or starch (2.5%) was added as adjuvant. The mixture was vigorously stirred to allow a homogenous dissolution of alginate. Then the mixture was extruded through sterile plastic nozzle with a diameter of 1mm and the resulting drops were then projected into sterile 6 g/1 CaCl<sub>2</sub> solution forming small calcium alginate matrix beads (2mm, mean diameter) entrapping bacterial cells. The beads were maintained in CaCl<sub>2</sub> solution at room temperature for additional 1-2h to obtain regular solid beads. The CaCl<sub>2</sub> solution was pumped out, and the beads were washed twice with sterilized distilled water. Fresh beads were either used directly as fresh, or kept at 4-5°C in sealed flacks for several days. Bacterial cells within 0.1g pellets were calculated after dissolving the pellets in phosphate buffer (pH7) solution by diluted agar plate.

Slow release of entrapped *Azospirillum brasilense* cells

A portion of 30 washed beads containing immobilized bacteria was transferred into 75 ml of sterile saline solution (0.1 % W/V NaCl) and gently shaken at 30°C for 24 h. Then triplicate samples of 0.5 ml of saline solution were collected, and the number of released bacteria was determined by the plate count method on nutrient agar plates. Then the pellets were rinsed twice in sterile water and transferred into a fresh saline solution, and the procedure was repeated after an additional 24 h up to 8 days.

#### **3.RESULTS AND DISCUSSION**

Slow release of A. *brasilense* ( $F_3$ ) cells from alginate beads enriched with various additives:

The release of *Azospirillum* cells from gel formulation was assessed up eight days. It was found that decrease of *Azospirillum* released from alginate beads from initial to  $8^{ht}$  day of screening. Among the various treatments, higher amount of *Azospirillum* cells (38.5x10<sup>4</sup> and 10.7x10<sup>4</sup> cfu/g bead) was released in Alginate + starch at initial and  $8^{th}$  day of sampling, respectively, followed by Alginate formulation with

skim

#### milk

addition

 $(29.0x10^4 - 15.0x10^3 \text{ cfu/g})$ . The lower amount of *Azospirillum* cells  $(27.5x10^4 - 13.1x10^3 \text{ cfu/g} \text{ bead})$  alginate alone gel formulation of initial and 8<sup>th</sup> day of sampling, respectively.

Entrapment of microbial cells has been reported to improve their metabolic activities and enhance the production of several hydrolyic enzymes (El-Katatny et al., 2004; El-Komy, 2005). Alginate immobilization has also been used as inoculant for plant growth promoting bacteria for over more than two decades (Bashan, 1986). The gel-like matrix allows the cells to remain viable and with its catalytic ability for longer duration (Bashan and Levanony, 1988 ;El-Komy, 2001). Moreover in the present study. The release of A. brasilense from alginate formulations with different additives (starch or skim milk) supported higher Azospirillum number released from the preparations when compared alginate formulations alone without any adjuvants. Sodium alginate with starch or skim milk might provide nutrients facilitating the bacterial multiplication within the bead. These findings are in accordance with many researches shown that the addition of adjuvants such as humic acid, starch, cellulose or skim milk is presumed to enhance the stability and provide protection and nutrition to the encapsulated microorganisms (Bashan, 1986, Shaban and El-Komy, 2000; El-Katatny et al., 2004, Kumaresan and Reetha, 2012).

Alginate beads with adjuvant	Initial	2days	4days	6days	8days
Alginate	27.5 X 10 <sup>4</sup>	18.0 X10 <sup>3</sup>	16.4 X10 <sup>3</sup>	16.1 X10 <sup>3</sup>	13.1X10 <sup>3</sup>
	(5.44)	(4.25)	(4.20)	(4.20)	(4.11)
Alginate +S.milk	29.0 X10 <sup>4</sup>	18.5 X10 <sup>3</sup>	18.5 X10 <sup>3</sup>	18.0X10 <sup>3</sup>	15.0 X10 <sup>3</sup>
	(5.46)	(4.27)	(4.26)	(4.25)	(4.18)
Alginate + Starch	38.5 X10 <sup>4</sup>	12.5 X10 <sup>4</sup>	12.0 X10 <sup>4</sup>	11.2 X10 <sup>4</sup>	10.7 X10 <sup>4</sup>
	(5.58)	(5.10)	(5.07)	(5.04)	(5.02)

Table .1: Slow release	of alginate	immobilized Az	ospirillum strain.
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Azospirillum population (CFU /g bead), and values in parenthesis are log No.



Fig.1: Slow release of alginate immobilized Azospirillum strain .

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